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Chemistry Matter Change Chapter 13

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13.78 kPa 6. Find the partial pressure of carbon dioxide in a gas mixture with a total pressure of 30.4 kPa if the partial pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa. 30.4 kPa 16.5 kPa 3.7 kPa 10.2 kPa Solutions Manual Chemistry: Matter and Change • Chapter 12 237

States of Matter

Solutions Manual Chemistry: Matter and Change • Chapter 13 255 CHAPTER 13 SOLUTIONS MANUAL Assume that the amount of gas is constant in the following problems 11 A sample of air in a syringe exerts a pressure of 102 atm at 220°C The APPENDIX D Problem Bank

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