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Ethanol in

Water Explain why  
ethanol ( $\text{CH}_3\text{CH}_2\text{OH}$ )

is more soluble in  
water than is ethane  
( $\text{CH}_3\text{CH}_3$ ). Answer

Ethanol is polar;  
ethane is not. The  
ethanol  $-\text{OH}$  group  
can hydrogen-bond  
with water. 2.

Calculation of pH from  
Hydrogen Ion

Concentration What is  
the pH of a solution  
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concentration of (a)  
1.75  $\times 10^{-5}$  mol/L; (b)

6.50  $\times 10^{-10}$  mol/L; (c)

1.0  $\times 10^{-4}$  mol/L;

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