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 $\text{KClO}_3$  decompose to form two formula  
units  $\text{KCl}$  and three molecules  $\text{O}_2$  b Four  
molecules  $\text{NH}_3$  react with

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answers the question. \_\_\_\_ 1. The first  
step in most stoichiometry problems is  
to \_\_\_\_.

- a. add the coefficients of the reagents
- c. convert given quantities to volumes
- b. convert given quantities to moles
- d ...

Chemistry Chapter 12

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Answer

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Chapter 12 Stoichiometry Test Answer

Key Answer:  $4.93 \times 10^{-5}$  L or  $49.3 \mu\text{L}$

In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent:  $[\text{Au}(\text{CN})_2]^-$ ,  $\text{LaCl}_3$ , ethanol, and para -nitrophenol. When

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Chapter 12 Multiple Choice Identify the  
letter of the choice that best completes  
the statement or answers the question.

\_\_\_\_ 1. The first step in most  
stoichiometry problems is to \_\_\_\_.

- a. add the coefficients of the reagents
- c. convert given quantities to volumes
- b. convert given quantities to moles
- d. convert given quantities to masses

\_\_\_\_ 2.

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