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Answer Key (continued) Chemistry: Matter and Change Supplemental Problems 5 12. Use the figure below to answer the following questions. a. How many valence electrons does an atom of this element have? 2 b. What is the atom's electron-dot structure? •Ca• c. If enough energy was added to remove an electron, from which energy level would the

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CHAPTER 5 Electrons in Atoms Chemistry: Matter and Change Supplemental Problems 7 1. Orange light has a frequency of 4.8 1014 s⁻¹. What is the energy of one quantum of orange light? up: 1s, 2s, 3s, 4s, 5s, 6s, 7s, 2p, 3p, 4p, 5p, 6p, 7p, 3d, 2. Which is greater, the energy of one photon of orange

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